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## ANTIOXIDANT ACTIVITY OF N-SUBSTITUTED İNDOLE 2- AND 3-CARBOXAMIDES

### N-SÜBSTİTÜE İNDOL-2 VE 3-KARBOKSAMİTLERİN ANTİOKSİDAN AKTİVİTESİ

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#### ABSTRACT

*A series of N-substituted indole amide derivatives was tested for in vitro effects on rat liver microsomal NADPH-dependent lipid peroxidation (LP) and superoxide anion (SOD) formation. Vitamin E (α-tocopherol) decreased the LP level by about 63% at 10<sup>-6</sup>M concentration. A significant decrease in male liver microsomal LP level was noted for the compounds 1, 8 and 10 at a concentration of 10<sup>-6</sup>M comparing with α-tocopherol. The inhibition of O<sub>2</sub><sup>-</sup> by compounds 8 and 10 were found highly efficient. Both activity results show that these compounds are promising antioxidants.*

**Key Words:** Antioxidant activity; N-substituted and N-H indole amide derivatives; superoxide dismutases; lipid peroxidation

#### ÖZET

*Bir seri N-süstitüe indol amid türevinin, sıçan karaciğerinden elde edilen mikrozomal NADP'ye bağımlı lipid peroksidasyon (LP) ve süperoksit anyon şekillenmesi üzerinde olan invitro etkileri test edildi. Vitamin E (α-tokoferol), LP seviyesini 10<sup>-6</sup> M konsantrasyonda yaklaşık % 63 kadar azalttı. Bileşiklerden 1,8 ve 10'ün 10<sup>-6</sup> M konsantrasyonda, α-tokoferol ile karşılaştırıldıklarında erkek karaciğer mikrozomal LP düzeylerinde belirgin bir azalmaya neden olduğu bulundu. Bileşiklerden 8 ve 10'ün, ö<sub>2</sub><sup>-</sup> anyonunu belirgin bir biçimde inhibe ettiği bulundu. Her iki aktivite sonucu, bu bileşiklerin ümit verici antioksidanlar olabileceklerini gösterdi.*

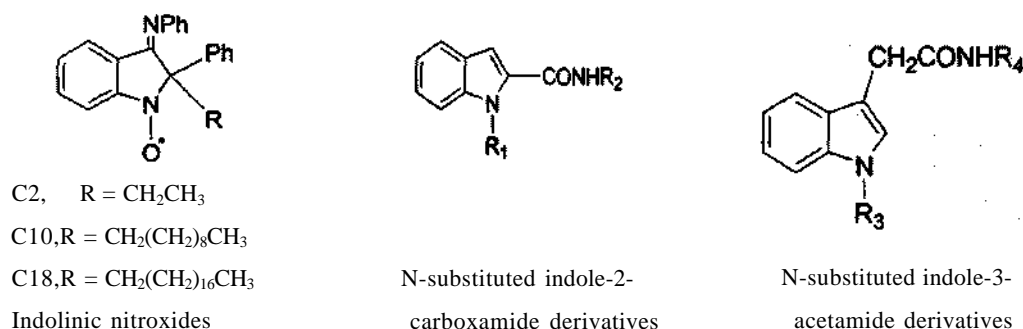
**Anahtar Kelimeler:** Antioksidan aktivite; N-süstitüe indole amit türevleri süperoksit dismutaz; lipid peroksidasyon

## INTRODUCTION

Indole structure was found noteworthy because it has been acknowledged as having interesting medicinal properties, such as antiinflammatory (1), antiallergic (2), antiviral (3, 4), anti-tumor (5,6), antimicrobial (6), antihypertensive activities (7). Recently, indole nitroxides (Figure 1) have also been reported to possess antioxidant properties (8).

Many non-steroidal antiinflammatory agents are known to act either by inhibiting the production of free radicals or by scavenging them (9). The relationship between inflammation and free radical's action is an important issue to design new antioxidant compounds, which can have good antiinflammatory activities.

In our previous article, we have described new some of N-substituted indole-2-carboxamide and indole-3-acetamide derivatives (Figure 1) with highly potent antioxidant properties (10). The structural similarity of previous antioxidant derivatives to novel N-substituted indole-2 and 3-carboxamide derivatives, which have been first synthesized for cyclooxygenase-2 (COX-2) inhibitors (11) by our laboratory, led us to evaluate of their antioxidant activities. Herein, we report the results of antioxidant properties and structure activity relationships between synthesized amide compounds.



**Figure 1:** Antioxidant indole compounds based on indole structure

## MATERIALS AND METHODS

Superoxide dismutase (SOD), cytochrome *c*, xanthine, thiobarbituric acid (TBA), malondialdehyde (MDA) were purchased from *Sigma* and potassium chloride, n-butanol, *o*-phosphorous acid were purchased from *Aldrich*.

### Cytocrome C Test Of Synthesized Compounds

Enzymatic formation of superoxide anion was assayed on the basis of cytochrome *c* reduction of as described by McCord and Fridovich (12). Superoxide anion ( $O_2^{\cdot-}$ ) was generated as a mixture of the enzyme xanthine oxidase and its substrate xanthine. The level of enzyme was detected by its ability to reduce cytochrome *c*, which causes arise in absorbance at 550 nm.

Enzymatic formation of superoxide anion was detected on the basis of cytochrome *c* reduction by xanthine oxidase plus xanthine. The incubation mixture (1.0 ml, total volume) consisted of phosphate buffer (pH= 7.8, 0.05M), xanthine (50 U.M), cytochrome *c* (60 mM), xanthine oxidase (0.32 units/ml) and different concentration of synthesized compounds at 100  $\mu$ M. The reaction was

started by addition of xanthine oxidase to this mixture and the reaction was conducted at 30 °C in a heating block. The cytochrome *c* reduction was monitored at 550 nm for 3 min. For any given set of condition, a calibration curve was obtained by using purified SOD from sigma as a standard.

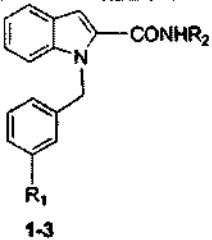
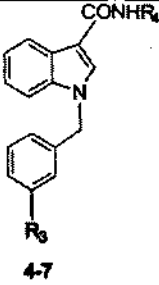
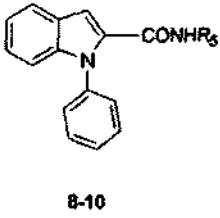








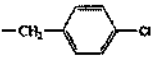

#### **Assay of lipid peroxidation**

The effect of different compounds on rat liver homogenate lipid peroxidation was determined by the method of Mihara et al. (13).

Lp Was Measured Spectrophotometrically For The Estimation Of Thiobarbituric Acid Reactant Substances (Tbars). Amount Of Tbars Were Expressed In Terms Of Nmol Malondialdehyde (Mda/G Tissue). Rat Liver Tissue Was Homogenized With 1.15% Kcl Solution To Make A 5% Homogenate. 0.5 ml Of Homogenate, 3 ml Of 1% O-Phosphorous Acid, 1 ml Of 0.6% Tba in Aqueous Solution And Various Concentrations Of Different Synthesized Compounds Were Added into A 10 ml Centrifuge Tube. The Mixture Was Heated For 45 Min in A Boiling Water Bath. After Cooling Down To Room Temperature, 4 ml Of N-Butanol Was Added And Mixed Vigorously. N-Butanol Phase Was Separated By Centrifugation And The Absorbance Was Measured At 532 And 520 nm. The Difference Was Determined As The Tba Value. Amounts Of Tbars Were Expressed in Terms Of Nmol Mda/G Tissue.

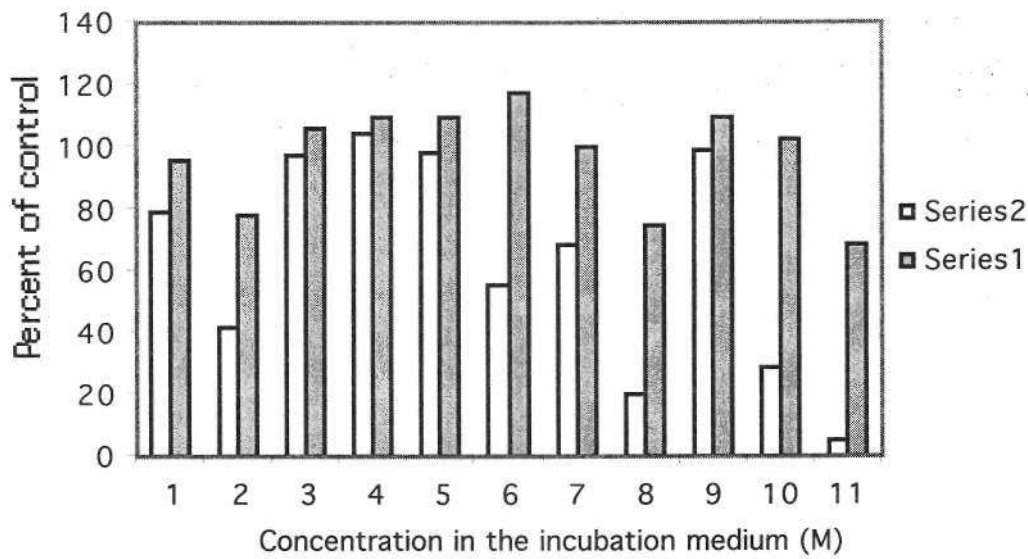
## RESULTS AND DISCUSSION

The Structure Features Of N-Substituted Indoleamide Derivatives 1-10 Were Shown In Table 1.

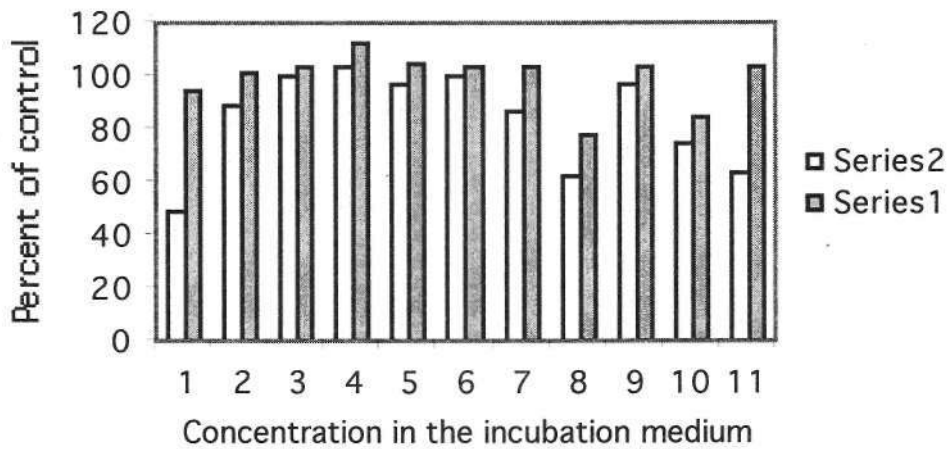
Table 1: The structural formula of indoleamide derivatives 1-10.							
 <p>1-3</p>			 <p>4-7</p>			 <p>8-10</p>	
Comp. No	R <sub>1</sub>	R <sub>2</sub>	Comp. No	R <sub>3</sub>	R <sub>4</sub>	Comp. No	R <sub>5</sub>
1	H		4	H		8	
2	H		5	F		9	
3	F		6	F		10	
			7	F			

These Derivatives Were Tested For *In Vitro* Effects On Rat Liver Microsomal NADPH-Dependent Lipid Peroxidation And Superoxide Anion Formation. The Comparative Concentration-Dependent Antioxidant Activities With  $\alpha$ -Tocopherol Were Shown In Figure 2 And Table 2.

Inhibition of superoxide production



Lipid peroxidation



a-tocopherol

**Figure 2:** Comparative Antioxidant Capacity Of Compounds 1-10 And cc-Tocopherol.

Among The Tested Compounds 8 And 10 Were Found Active At 10<sup>-4</sup> M Concentration When Compared With The Others. Moreover, These Compounds Showed Similar Activities Against 30 Iu Superoxide Dismutase. Therefore, inhibition Of O<sub>2</sub><sup>•-</sup> By 8 And 10 Are Likely To Render These Compounds As Promising Antioxidant. The Activity Results Show That N-

Antioxidant Properties. This Clearly indicates That The Phenyl Ring Might Be Essential For The Scavenging Of Free Radicals And influencing The Activity. The Activity Results Show The Fact That These Compounds Are Potent Antiinflammatory Agent.

In This Assay, Free Radicals Are Generated Toward Carbon-Center Compounds By Photo-Oxidation, Which Was Explained in Our Previous Article (10). This Assay Can Be Used To Determine Whether A Compound is A General Free Radical Scavenger Or Scavenger Specific For The Superoxide Anion. A Substance With No Free Radical Scavenging Activity Does Not Have Any Effect On The Assay (14).

As Can Be Seen From Table 2 And Figure 2, Compounds 1, 8 And 10 Showed inhibition Of Lp Levels For The Concentration Level At  $10^{-4}$ m.

While Compound 1 Have inhibition On Lp Level, it Does Not Have Any Effect On Superoxide Anion Production. This Difference is Not Surprising Since The Mechanism Of Production Of Reactive Oxygen Species is Different in These Assays.

**Table 2:** Effects of the compounds **1-10** on liver LP levels and liver superoxide anion production

Compound	Concentration in the incubation medium (M)	superoxide anion production percent of control <sup>1</sup>	LP (nmol MDA/g tissue) <sup>a</sup>	Percent of control
<b>1</b>	$10^{-4}$	79±3.5	15.65±0.7	49
	$10^{-5}$	96±7.0	30.28±1.6	94
<b>2</b>	$10^{-4}$	42±3.4	28.42±0.9	89
	$10^{-5}$	78±1.4	32.58±0.4	101
<b>3</b>	$10^{-4}$	97±7.0	32.26±0.8	100
	$10^{-5}$	106±2.1	33.24±2.7	103
<b>4</b>	$10^{-4}$	104±7.0	33.15±1.5	103
	$10^{-5}$	110±6.3	36.13±0.8	112
<b>5</b>	$10^{-4}$	98±0.7	31.36±1.1	97
	$10^{-5}$	110±5.4	33.55±2.7	104
<b>6</b>	$10^{-4}$	56±2.8	32.26±1.2	100
	$10^{-5}$	117±3.4	33.14±1.0	103
<b>7</b>	$10^{-4}$	69±5.6	28.18±1.7	87
	$10^{-5}$	100±1.4	33.22±1.5	103
<b>8</b>	$10^{-4}$	20±2.8	20.12±1.8	62
	$10^{-5}$	75±5.6	25.12±0.9	78
<b>9</b>	$10^{-4}$	99±1.4	31.36±0.9	97
	$10^{-5}$	110±6.3	33.14±0.4	103
<b>10</b>	$10^{-4}$	29±5.6	23.87±0.9	74
	$10^{-5}$	103±3.5	27.10±0.5	84
<b>Control<sup>b</sup></b>			32.26±1.8	100
<b>a-tocopherol</b>	$10^{-4}$	5±2.3	20.50±2.4	63
	$10^{-5}$	69±5.1	33.14±1.9	103
<b>Control<sup>c</sup></b>		100		
<b>SOD</b>	30 IU	28±2.7		
	45 IU	13±1.2		

<sup>a</sup>Each value represents the mean + S.D. of 2-4 independent experiments.

<sup>b</sup> Methanol plus dimethylsulfoxide ; control solvent for compounds 1-10.

\* water; control for SOD.

### Acknowledgments

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